## SOLANOLIDE, A STEROID LACTONE SAPOGENIN FROM SOLANUM HISPIDUM\*

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**Key Word Index**—Solanum hispidum; Solanaceae; sapogenin;  $3\beta$ ,  $6\alpha$ ,  $16\beta$ -trihydroxy- $5\alpha$ -pregnane-20S-carboxylic acid (22, 16)-lactone; <sup>1</sup>H NMR; <sup>13</sup>C NMR.

**Abstract**—Solanolide, a new  $C_{22}$  steroid lactone sapogenin isolated from the leaves of *Solanum hispidum* Pers., has been characterized as  $3\beta$ ,  $6\alpha$ ,  $16\beta$ -trihydroxy- $5\alpha$ -pregnane-20S-carboxylic acid (22, 16)-lactone from <sup>1</sup>H and <sup>13</sup>C NMR analyses and correlation with neochlorogenin.

Recently we have reported the isolation of a number of new spirostane saponins and sapogenins from the leaves [1, 2] and seeds [3] of Solanum hispidum. We now report another new C<sub>22</sub> steroid sapogenin, designated as solanolide, from the leaves of this plant.

Solanolide (1),  $C_{22}H_{34}O_4$  (M<sup>+</sup> at m/z 362), mp 252-254°,  $[\alpha]_D - 28^\circ$  (CHCl<sub>3</sub>; c 0.5), was isolated as fine needles (MeCN) upon acidification of the alkalisoluble part of the crude aglycone mixture [1] obtained by Smith degradation of the major dirhamnoside frac-

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tion of the leaf extract. The yield was very poor (1.5%) based on the crude aglycone). It could not be methylated with MeOH-HCl or with Me<sub>2</sub>SO<sub>4</sub>-NaOH indicating the absence of any free phenolic hydroxyl or carboxyl group. That it was a lactone became evident from its IR spectrum (KCl) which showed a strong absorption band at 1760 cm<sup>-1</sup> for a five-membered lactone, besides a broad band between 3500 and 3100 cm<sup>-1</sup> for hydroxy group(s). Its mass spectrum exhibited intense peaks at m/z 344  $[M-H_2O]^+$ , 329  $[344-Me^{-}]^{+}$ , 326  $[344-H_{2}O]^{+}$ , 311  $[326-Me^{-}]^{+}$ , 289  $[M - C_3H_5O_2^*]^+$ , 271  $[289 - H_2O]^+$ , 264  $[M - C_5H_6O_2^*]^+$ 246  $[M - H_2O - C_5H_6O_2]^+$ , 231  $[246 - Me^{-}]^+$  and 213 [231 – H<sub>2</sub>O]<sup>+</sup> indicating the presence of two hydroxyl groups in the molecule. The remaining two oxygen atoms form part of the lactone ring. The location of the

a m/z 289

b m/z 264

Compound	19-Me	18-Me	21-Me	16-H	20-H	3-Н	6-H	OAc
1	0.84	0.75	1.31 d	4.98 ddd	2.59 q	3.52 m	3.52 m	
			(7)	(7, 7, 4)	(7)			
1†	0.82	0.71	1.24 d	4.90 ddd	$2.65 \ q$	3.86 m	3.63 ddd	
			(7)	(7, 7, 4)	(7)		(10, 10, 4)	
2	0.91	0.75	1.31 d	4.97 ddd	2.58 q	4.68 m	4.68 m	2.02
			(7)	(7, 7, 4)	(7)			
3	0.96	0.78	1.31 d	4.94 ddd				
			(7)	(7, 7, 4)				

Table 1. <sup>1</sup>H NMR spectral data (δ) of solanolide (1) and its derivatives\*

Table 2. <sup>13</sup>C Chemical shifts\* of solanolide (1) and its derivatives

Carbon	1.5	2.5	2+	<b>5</b> §
no.	1†	2‡	3‡	28
1	38.2	36.8	33.0	36.8
2	32.1	27.0	37.1	27.1
3	70.8	72.8	210.5	73.0
4	33.2	28.2	36.9	28.4
5	52.5	48.4	57.4	48.5
6	68.3	71.6	207.7	72.0
7	42.5	37.6	46.3	37.8
8	33.9	33.4	37.2	33.7
9	54.2	53.4	53.4	53.6
10	36.5	36.6	41.0	36.6
11	20.8	20.4	21.0	20.9
12	37.9	38.0	37.7	39.8
13	41.7	41.7	42.1	40.5
14	54.4	54.1	54.6	55.9
15	33.5	32.9	32.8	31.6
16	82.6	82.3	82.1	80.6
17	58.9	58.8	58.8	61.9
18	13.8	13.7	13.8	16.4
19	13.6	13.2	12.6	13.3
20	36.2	36.0	36.0	42.1
21	17.9	17.8	17.9	14.3
22	180.9	180.8	180.7	109.6
<b>COMe</b>		170.3		170.5
		170.5		170.7
COCH <sub>3</sub>		21.1		21.3
		21.3		21.4

<sup>\*</sup>Spectra were recorded in a Jeol FX-100 spectrometer at 25.05 MHz. Conditions for measurements were: data points, 8K; spectral width, 6 KHz, repetition time, 1.0 or 1.2 sec; flip angle, 25-40°. The chemical shifts expressed on the  $\delta$  scale using TMS as internal standard. The assignments were based on the SFORD spectra, low-power noise-modulated decoupled spectra and selective proton decoupled spectra (where applicable) as well as the chemical shifts of  $\delta \alpha$ -spirostanols and their acetates [10].

lactone chromophore in the terminal E ring was evident from the formation of ions at m/z 289 (species a) and 264 (species b) by the expulsion of  $C_3H_5O_2$  and  $C_5H_6O_2$  radicals respectively from the molecular ion. The peak at m/z 264 appears to be diagnostic for the lactones of the solanolide type since the peak at m/z 289 could also be obtained from the spirostane derivatives with two hydroxy groups in the androstane moiety [1, 2].

Acetylation of 1 with  $Ac_2O$ -pyridine at room temperature afforded a diacetate (2), mp 250–252°,  $[\alpha]_D$  – 46° (CHCl<sub>3</sub>; c 0.5),  $IR \nu_{\rm max}^{\rm KCl}$  cm<sup>-1</sup>: 1770 (lactone), 1725 and 1250 (acetate); m/z 446 [M]<sup>+</sup>, 386 [M – AcOH]<sup>+</sup>, 371 [386 – Me<sup>-</sup>]<sup>+</sup>, 344 [386 –  $C_2H_2O$ ]<sup>+</sup>, 326 [386 – AcOH]<sup>+</sup> and 311 [326 – Me<sup>-</sup>]<sup>+</sup>. On oxidation with Kiliani reagent, 1 yielded a diketone (3), mp 248–250°,  $[\alpha]_D$  – 45° (CHCl<sub>3</sub>; c 0.4),  $IR \nu_{\rm max}^{\rm KCl}$  cm<sup>-1</sup>: 1760 (lactone), 1700 (six-membered C=O); m/z 358 [M]<sup>+</sup>, 343, 329 and 205, indicating that both the hydroxy groups of 1 are secondary ones and are located in six-membered rings.

The <sup>1</sup>H NMR chemical shifts (Table 1) of the C-19 methyl protons of solanolide (1) and its acetate (2) were found to be very close to those of a co-occurring sapogenin, viz. neochlorogenin (4) and its acetate (5)[1], respectively. Since the chemical shifts of angular methyl protons are dependent [4, 5] on the position and orientation of the hydroxyl group in the androstane moiety, it can be presumed that solanolide possesses the same  $3\beta$ ,  $6\alpha$ -dihydroxy- $5\alpha$ -androstane ring system which was also supported by the multiplicity of the signals for H-3 (broad multiplet whose  $W_{1/2}$  could not be measured correctly) and H-6 (ddd, J = 10, 10, 4 Hz). The presence of only one secondary methyl group resonating at somewhat lower field ( $\delta$ 1.31, J = 7 Hz) indicated that Me-21 must be adjacent to the lactone carbonyl. Additional support for the substitution pattern in the A/B ring was also obtained from the spectrum of 3 which showed the expected chemical shift ( $\delta$  0.96) for the C-19 methyl group of such a 3, 6-diketo- $5\alpha$ -steroid[1, 2].

The above conclusions were also borne out by the  $^{13}$ C NMR spectra (Table 2) of solanolide (1) and its derivatives (2 and 3). Thus, the chemical shifts for the A and B ring carbons of solanolide acetate (2) were in excellent agreement with the values found [10] for the  $3\beta$ ,  $6\alpha$ -diacetoxy-spirostane sapogenin (5). The

<sup>\*</sup>Spectra were recorded in CDCl<sub>3</sub>. The values in parentheses are the coupling constants in Hz.

<sup>†</sup>Spectrum recorded in pyridine- $d_5$ .

<sup>†</sup>Spectrum recorded in pyridine-d<sub>5</sub>.

<sup>‡</sup>Spectra taken in CDCl<sub>3</sub>.

<sup>§</sup>Chemical shifts included from Ref. [10].

The closely lying peaks may be interchanged.

chemical shifts for the same carbons of the diketone (3) also corresponded with the calculated [11] values (C-1, 38.2; C-2, 37.3; C-4, 35.8; C-5, 58.1; C-7, 46.6; C-8, 38.0; C-9, 54.1 and C-10, 41.1) within  $\pm 1$  ppm, a deviation in reasonable agreement with the proximity of the two carbonyl functions. On the other hand, considerable differences in the resonance frequencies of C-15-C-17 and C-20-C-22 of 1-3 compared to those of 5 reflected the location of the lactone chromophore in the E ring.

Finally, the structure of the compound was established as  $3\beta$ ,  $6\alpha$ ,  $16\beta$ -trihydroxy- $5\alpha$ -pregnane-20S-carboxylic acid (22, 16)-lactone (1) by its correlation with neochlorogenin (4). Thus, neochlorogenin acetate (5) on oxidation with nitric acid [6] in Et<sub>2</sub>O-CHCl<sub>3</sub> (4:1) at room temperature yielded a product exactly identical (mp, mmp, IR, TLC) with solanolide (1).

Only two other lactones of this type have so far been reported [6-9] from other Solanum species.

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